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Classical thermodynamics considers three main kinds of thermodynamic process: change in a system, cycles in a system, and flow processes. Defined by change in a system, a thermodynamic process is a passage of a thermodynamic system from an initial to a final state of thermodynamic equilibrium. The initial and final states are the defining elements of the process.

**Thermodynamic process - Wikipedia**

The Biot–Savart law is used for computing the resultant magnetic field  $B$  at position  $r$  in 3D-space generated by a steady current  $I$  (for example due to a wire). A steady (or stationary) current is a continual flow of charges which does not change with time and the charge neither accumulates nor depletes at any point. The law is a physical example of a line integral, being evaluated over the ...

**Biot–Savart law - Wikipedia**

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